

# Development of a novel thermoelectric module based device for thermal stability measurement of phase change materials

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## Development of a novel thermoelectric module based device for thermal stability measurement of phase change materials

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### Abstract

A recently developed method for the thermal stability measurement of phase-change materials (PCMs) involves thermal cycling using a thermoelectric module as a heating and cooling element. However, the utility of this approach was found to have some limitations, mainly because the thermoelectric polarity is changed according to time rather than the actual sample temperature. A method for thermal cycling test, where the thermoelectric polarity is automatically changed according to the sample temperature was developed in this study. In addition, a new cartridge design in this device requires a small sample volume (1.53 cm<sup>3</sup>) and can be easily assembled and disassembled. This proposed device was tested on beeswax as a PCM sample. This is very important for savings PCMs material which usually expensive. The results showed that the apparatus had automatically cycled between the melting and cooling temperatures of beeswax. The thermal data showed that beeswax retains consistent melting and freezing temperatures after 1000 cycles, however, its heat of fusion degrades over repeated thermal cycling. This apparatus can be readily applied to study a wide range of PCMs for such as thermal energy storage materials for energy conservation. To our best knowledge, yet no study has been performed on this kind of equipment so far.

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### Introduction

Thermal energy storage (TES) has been widely studied to meet our daily energy needs and to reduce the consumption of fossil fuels [1], and it can also be leveraged for thermal heating and cooling. In the domestic

and industrial sectors, including superior storage efficiency, phase-change materials (PCMs) offer several advantages for TES applications [2,3]. For instance, PCMs store energy by transforming between the solid and liquid phases while absorbing or releasing heat. A high thermal conductivity of PCMs is desirable for TES applications to facilitate rapid heat distribution and absorption. PCMs can be grouped into organic, inorganic, and eutectic materials [4], each of which offers certain advantages and disadvantages. Organic materials provide excellent thermal resistance, nontoxicity, and low sub-cooling potential, but they have poor thermal conductivity and low latent volumetric capacity [1].

Nanoparticles have been used to increase the thermal conductivities of PCMs and to reduce the sub-cooling potential, but they also decrease the latent heat capacity [5,6]. All the PCMs provide good calorific storage efficiency. The other unique thermal characteristics of PCMs, such as conductivity, melting and freezing temperatures, and heat storage capacity are important factors for applications, including air conditioning systems, solar energy systems, and cooling systems for electronics [[7], [8], [9], [10]]. Unfortunately, the thermal characteristics of PCMs tend to degrade over a certain period of time. This characteristic is very important for commercial PCM. Thus, there is considerable interest in developing PCMs that provide consistent performance or without an excessive change in its thermos-physical and chemical properties over a long period of time [4]. That performance is called thermal stability and reliability. A period of charging and discharging of PCM or thermal cycling of PCM can happen one or more times in a 24 h time period, implying at least a 1000 cycle for a 3 years life of a PCM storage [11].

There is a need for a device to measure the thermal stability and reliability of a PCM over repeated cycles of heating and cooling. Generally, thermal cycling test devices utilize specific measuring instruments for heating and cooling. However, there is no standardized device for testing PCMs under thermal cycling [12]. Build-up methods have been proposed in several studies to test the thermal stabilities of PCMs. For example, Sari et al. used a thermostatic bath of a fatty acid-based PCM by heating it with a heating controller until it reached the melting point and then switching off the controller to allow the sample to cool to room temperature [[13], [14], [15]]. Sharma et al. used an electric hot plate to heat the PCM sample inside a stainless steel container followed by cooling to room temperature [[16], [17], [18]]. Shukha et al. introduced an oven to heat the PCM to a fixed temperature and removed the sample from the oven to allow it to cool to room temperature [19]. Silakhori et al. fabricated a copper box, filled it with the PCM, and attached a strip steel heater to one side and a heat pipe, heat sink, and cooling fan to the other side [20]. Sharma and Shukla used two chambers filled with water to heat and cool the PCM samples [21]. These previous studies conducted the thermal cycle always 1000 cycles and lower than 1000 cycles.

Recently [11], have measured the thermal cycle of organic binary PCM. They developed a new tailor-made thermal cycling apparatus, which consist of heating and cooling sources, water pump, the whole water tube network, water tank, and two solenoid adjusting valves. Three types of composite fatty acids Capric acid (CA) & Lauric acid (LA), CA & Myristic acid (MA), and CA & Palmitic acid (PA), were tested with several mass ratios. The thermal cycling number is up to 4000 and DSC tests are carried out after each testing stage.

Lately, we have proposed a new method, employing pouring the PCM (in this case, RT22HC) into a cartridge as a test cell and attaching thermoelectric elements to both sides of the cartridge for heating and cooling the sample [22]. With this setup, we demonstrated the PCM could undergo heating and cooling process up to 1000 cycles [22]. However, in this method, the change in the polarity of the thermoelectric elements, which was controlled by a programmable DC power supply, was triggered only by the elapsed time according to

the predicted time required for the PCM to melt and solidify. Moreover, the cartridge requires a large sample volume, makes it difficult to remove the sample after the test, and it must be cleaned after every single use. Thus, the research aimed to develop a thermal cycling test method using a thermoelectric module that can change its polarity based on the sample temperature and cartridge, and the module can be easily dismantled.

In order to test the proposed device, the thermal cycling tests on beeswax was performed as an example of a PCM and characterized its thermal properties by differential scanning calorimetry (DSC). The main contribution of this study is an integrated cartridge design development and cycling testing for PCM as an advantage from the previous method. To our best knowledge, yet no study has been performed on the development of an integrated cartridge design development and cycling testing for PCM application so far.

## Section snippets

### Cartridge design

The disadvantage of the previous method was that a large amount of the PCM sample had to be poured into the cartridge because of its large dimensions. The excess PCM sample requires significant costs when the raw PCM is expensive and not widely available in the market. Thus, a modified cartridge was developed for the thermal cycling test, as shown in Fig. 1. In this case, the cartridge was sandwiched between thermoelectric modules on the left and right sides to provide simplicity and...

### Beeswax thermal cycling test process

Fig. 6 shows the result of the beeswax thermal cycling test using the modified method. The results show that the newly proposed device successfully changed the polarity to alternately heated and cooled the beeswax sample automatically. The heating and cooling processes were adjusted during the operation based on the melting and freezing points of the beeswax sample.

The heating process lasted for 3 min, and the cooling process took 4 min; thus, each cycle for the beeswax sample was approximately ...

## Conclusions

The proposed new device using thermoelectric modules was demonstrated to be an effective alternative for PCM thermal cycling tests. The device is automatically switched the thermoelectric polarity based on the sample temperature. The heating process was tested on a beeswax sample, and the results showed that the sample reached its melting point and was heated to 65 °C during the heating process. Moreover, the cooling process continued until the sample reached its freezing point at 46 °C. The...

## Acknowledgments

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2021, Materials Science and Engineering B: Solid-State Materials for Advanced Technology

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...In recent years, various types of TE materials have been studied. For example, double-filled p-type skutterudites [5,6], layered cobalt oxides [7,8], half-Heusler alloys [9–11], superlattices [12–15], composite films [16–18], phase change materials [19–20], and so on. In the last few years, Sb<sub>2</sub>(S,Te)<sub>3</sub> has been synthesized into a variety of nanostructured forms, which exhibits reinforced electronic properties to obtain optimal ZT [46–48]...

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2020, Case Studies in Thermal Engineering

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...The research quite encouraging but did not give an optimized result and therefore Amin et al. [10] attempt to analyse a new material from beeswax/graphene thermal as PCM. This research then following by using beeswax/multi-walled carbon nanotubes and attempted to used novel thermoelectric module based devices for thermal stability measurement of PCM [11,12]. One of the reasons for global warming is the increasing number of vehicle owners each year, which has consequences for emissions and fuel availability [13]...

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